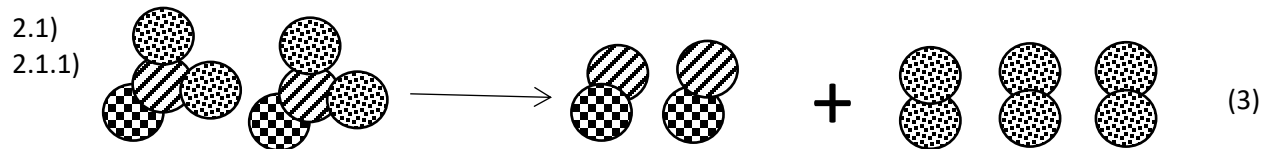


**Chemiese
Reaksies
Toets
Memorandum**

VRAAG 1:

- 1.1) B.
- 1.2) B.
- 1.3) C.
- 1.4) C.
- 1.5) A.

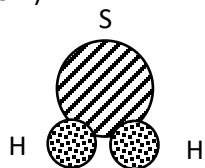
VRAAG 2:



- 2.1.2) KCl en O₂. (2)
- 2.1.3) Kalium, Chloor, Suurstof. (3)
- 2.1.4) 12 atome. (1)

VRAAG 3:

- 3.1)
- 3.1.1) Waterstof as 'n reaktant.
- 3.1.2) Waterstofsulfied as 'n produk. (2)
- 3.2) Links van die pyl word waterstof voorgestel as diatomies en regs van die pyl is waterstof in 'n verbinding dus is waterstof deel van die molekule. (2)
- 3.3) Verbinding. (1)
- 3.4)



(3)